

## Synthesis and Characterisation of Nona-osmium Carbonyl Clusters; Crystal and Molecular Structure of $[(\text{Ph}_3\text{P})_2\text{N}] [\text{Os}_9\text{H}(\text{CO})_{24}]^-$

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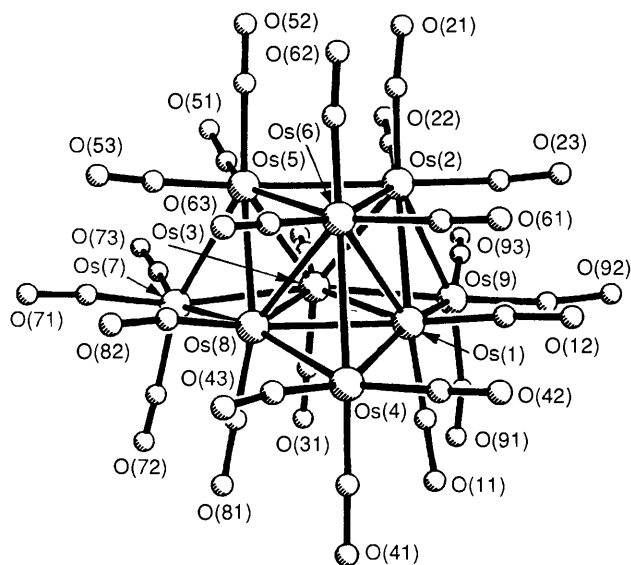
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The anion  $[\text{Os}_9\text{H}(\text{CO})_{24}]^-$ , **2** has been isolated in moderate yield by the vacuum pyrolysis of  $[\text{Os}_3(\text{CO})_{10}(\text{NCMe})_2]$ , and has been shown by a single crystal X-ray structure analysis to contain the previously unobserved tricapped octahedral  $\text{Os}_9$  metal core.

Extensive studies have been carried out on the structure and reactivity of high nuclearity osmium cluster compounds.<sup>1-4</sup> However, nona-osmium carbonyl clusters are extremely rare. The only report<sup>5</sup> of  $\text{Os}_9$  species is the isolation of

$[\text{Os}_9(\text{CO})_{21}\{\text{CHC}(\text{R})\text{CH}\}]^-$  (R = Me or Et) from the thermolysis of  $[\text{Os}_3(\text{CO})_{12}]$  in isobutyl alcohol or 2-methylbutan-1-ol, in very low yield (less than 2%).

We now report a synthetic route to a series of new nona-osmium species  $[\text{Os}_9(\text{H})_2(\text{CO})_{24}]$  **1**,  $[\text{Os}_9\text{H}(\text{CO})_{24}]^-$  **2** and  $[\text{Os}_9(\text{CO})_{24}]^{2-}$  **3**. Vacuum pyrolysis of  $[\text{Os}_3(\text{CO})_{10}(\text{NCMe})_2]$  at 170 °C for 16 h gives a dark brown microcrystalline solid. The solid mixture of clusters is first extracted with  $\text{CH}_2\text{Cl}_2$  to remove  $[\text{Os}_5(\text{CO})_{16}]$  (3%),  $[\text{Os}_6(\text{CO})_{18}]$  (18%),  $[\text{Os}_7(\text{CO})_{21}]$  (8%) and  $[\text{Os}_8(\text{CO})_{23}]$  (trace). The residue is then extracted with acetone to give a dark brown solution and a yellow powder of  $[\text{Os}_3(\text{CO})_{12}]$  (22%). Separation of the brown solution by TLC on silica (eluent; acetone-hexane 1 : 1) gives the new  $[\text{Os}_9\text{H}(\text{CO})_{24}]^-$  **2** mono-anion ( $R_f \sim 0.6$ ) in 20% yield. The stoichiometry of **2** was initially established by FAB MS<sup>†</sup> and <sup>1</sup>H NMR and has been confirmed by a single crystal X-ray structure analysis of the  $[(\text{Ph}_3\text{P})_2\text{N}]^+$  salt.<sup>‡</sup> The molecular structure of **2**, together



**Fig. 1.** The molecular structure of the  $[\text{Os}_9\text{H}(\text{CO})_{24}]^-$  **2** anion showing the atom numbering scheme. Principal bond distances (Å): Os(1)–Os(2), 2.815(3); Os(1)–Os(3), 2.892(3); Os(1)–Os(4), 2.738(3); Os(1)–Os(6), 2.816(3); Os(1)–Os(8), 2.886(3); Os(1)–Os(9), 2.735(3); Os(2)–Os(3), 2.831(3); Os(2)–Os(5), 2.924(3); Os(2)–Os(6), 2.942(3); Os(2)–Os(9), 2.938(3); Os(3)–Os(5), 2.822(3); Os(3)–Os(7), 2.739(3); Os(3)–Os(8), 2.874(3); Os(3)–Os(9), 2.745(3); Os(4)–Os(6), 2.960(3); Os(4)–Os(8), 2.729(3); Os(5)–Os(6), 2.946(3); Os(5)–Os(7), 2.936(3); Os(5)–Os(8), 2.813(3); Os(6)–Os(8), 2.823(3); Os(7)–Os(8), 2.741(3).

<sup>†</sup> Spectroscopic data for **1**: IR  $\nu(\text{CO})/\text{cm}^{-1}$  ( $\text{CH}_2\text{Cl}_2$ ) 2096w, 2087m, 2077s, 2070sh, 2042w and 2026w. For the  $[(\text{Ph}_3\text{P})_2\text{N}]^+$  salt of **2**: IR  $\nu(\text{CO})/\text{cm}^{-1}$  ( $\text{CH}_2\text{Cl}_2$ ) 2057s, 2044s, 2020m, 2001m and 1990w sh; FAB MS  $M^+$  (obs.)  $m/z$  2385 (calc.) 2385; <sup>1</sup>H NMR ( $\text{CD}_2\text{Cl}_2$ ,  $\delta$  –9.2 (s, 1H, MH) and 7.62 (m, 30H,  $[(\text{Ph}_3\text{P})_2\text{N}]^+$ ). For the  $[(\text{Ph}_3\text{P})_2\text{N}]^+$  salt of **3**: IR  $\nu(\text{CO})/\text{cm}^{-1}$  ( $\text{CH}_2\text{Cl}_2$ ) 2035s, 2023s, 1998m, 1975m and 1965w, sh; FAB MS  $M^+$  (obs.)  $m/z$  2384 (calc.) 2384.

<sup>‡</sup> Crystal data for **2**:  $\text{C}_{60}\text{H}_{31}\text{NO}_{24}\text{P}_2\text{Os}_9$ ,  $M = 2923.6$ , monoclinic, space group  $P2_1/n$  (non-standard setting of  $P2_1/c$  No. 14),  $a = 12.001(2)$ ,  $b = 31.339(10)$ ,  $c = 17.602(3)$  Å,  $\beta = 90.57(2)^\circ$ ,  $U = 6620(3)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 2.933$  g cm<sup>-3</sup>,  $F(000) = 5216$ , Mo-K $\alpha$  radiation,  $\lambda = 0.71069$  Å,  $\mu(\text{Mo-K}\alpha) = 173.52$  cm<sup>-1</sup>, 3770 observed diffractometer data [ $F > 3\sigma(F)$ ]. Structure solved by direct methods and Fourier difference techniques, refined by full-matrix least-squares analysis (Os and P anisotropic) to  $R = 0.069$ ,  $R_w = 0.063$ . Atomic coordinates, bond lengths and angles, and thermal parameters for both compounds **2** and **3** have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

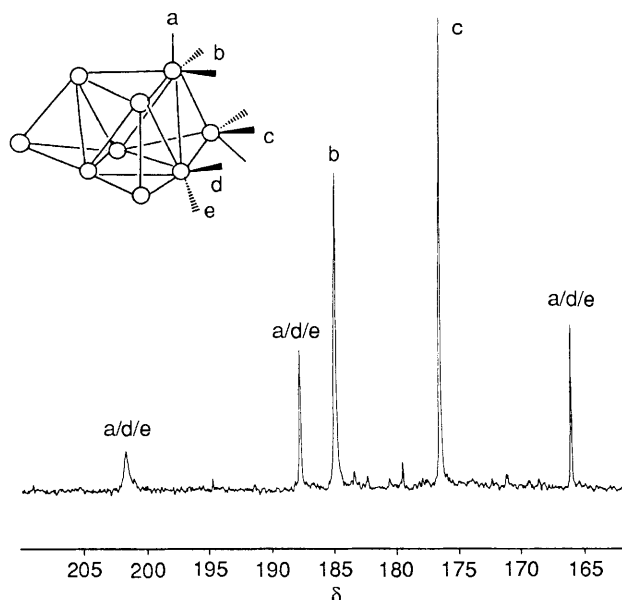


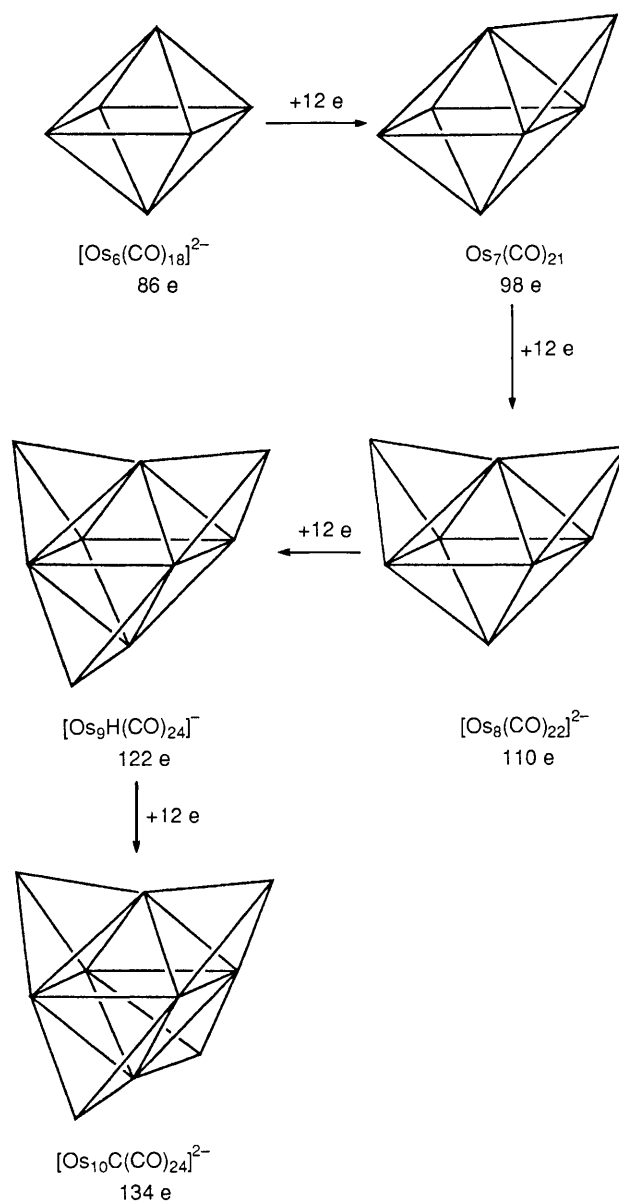
Fig. 2  $^{13}\text{C}\{^1\text{H}\}$  NMR spectrum of 50%  $^{13}\text{C}$  enriched  $[(\text{Ph}_3\text{P})_2\text{N}]^+[\text{Os}_9\text{H}(\text{CO})_{24}]^-$  ( $\text{CD}_2\text{Cl}_2$ , 298 K)

with some important bond parameters is shown in Fig. 1. The metal core can be described as a tricapped octahedron and may be viewed as being derived from the  $\text{Os}_{10}$  framework in  $[\text{Os}_{10}\text{H}_4(\text{CO})_{24}]^{2-}$  by the removal of one cap.<sup>6</sup> The capping  $\text{Os}(\text{CO})_3$  groups are asymmetric with one long and two short bonds [2.940(10) and 2.739(5) Å] to the central octahedron so that the atoms Os(4), Os(7) and Os(9) lie below the plane defined by Os(1), Os(3) and Os(8) with average deviation 0.30 Å from the plane. This feature is not observed in  $[\text{Os}_{10}\text{H}_4(\text{CO})_{24}]^{2-}$  where the four  $\text{Os}_6$  frameworks, defining the tetrahedron faces, are planar.<sup>6</sup> The mono-anion **2** exhibits an approximate  $C_{3v}$  symmetry. All carbonyl ligands are terminally bonded and essentially linear. The position of the hydride could not be determined directly by X-ray analysis, but potential energy calculations<sup>7</sup> suggest that the hydride atom  $\mu_3$ -bridges Os(1), Os(3), Os(8) face. At 25°C, the  $^{13}\text{C}\{^1\text{H}\}$  NMR spectrum shows five resonances in the region  $\delta$  205–165 with intensity ratio 1:1:2:3:1 which is consistent with the solid state structure assuming the turnstile rotation of the carbonyls on Os(2), Os(5) and Os(6) is restricted, see Fig. 2.

Reaction of **2** with DBU (1,8-diazabicyclo[5.4.0]undec-7-ene) gives the dianion **3** in quantitative yield which can be isolated as  $[(\text{Ph}_3\text{P})_2\text{N}]^+$ ,  $[\text{Ph}_4\text{P}]^+$  or  $[\text{Bu}_4\text{P}]^+$  salts. A preliminary X-ray analysis<sup>8</sup> of  $[\text{Ph}_4\text{P}]^+$  salt of **3** reveals an identical metal core to that found in **2**. However, the poor crystal quality and severe absorption effect of the crystal precluded the determination of ligand positions. Protonation of **3** with  $\text{CF}_3\text{CO}_2\text{H}$  in  $\text{CH}_2\text{Cl}_2$  gives **2** and further protonation can be achieved by  $\text{HBF}_4$  to give **1** which dissociates easily back to **2** in  $\text{CH}_2\text{Cl}_2$ .

Both **2** and **3** contain 122 valence electrons and the observed structures represent the first examples of the simple tricapped octahedral geometry of  $M_9$ -cluster. In the light of electron counting and structural relation of transition metal clusters, they fit in the capping principle based on the Wade<sup>8</sup> and Mingos<sup>9</sup> approaches, see Scheme 1.

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Scheme 1 The relationship between structure and the total polyhedral electron count for osmium cluster species

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§ Crystal data for **3**:  $\text{C}_{72}\text{H}_{40}\text{O}_{24}\text{P}_2\text{Os}_9$ ,  $M = 3062.9$ , monoclinic, space group  $P2_1/n$  (non-standard setting of  $P2_1/c$  No. 14),  $a = 13.168(9)$ ,  $b = 20.473(5)$ ,  $c = 27.033(9)$  Å,  $\beta = 94.46(4)^\circ$ ,  $U = 7265.7$  Å<sup>3</sup>,  $Z = 4$ , 2351 observed diffractometer data [ $F > 3\sigma(F)$ ] with current  $R$  factor 0.198.